

LAMPIRAN

Lampiran 1. Perhitungan Sampel Dan Pelarut Untuk Ekstraksi

Ekstraksi dengan metode maserasi menggunakan perbandingan

sampel : pelarut = 1 : 3

= 250 gram ampas teh hitam : 750 ml aquadest

Lampiran 2. Perhitungan Rendemen

Rendemen ekstrak ampas teh hitam (*Camellia sinensis*).

- a. Berat sampel ampas teh hitam = 250 gram
- b. Berat beaker glass kosong (a) = 157 gram
- c. Berat beaker glass + ekstrak cair (b) = 376
- d. Berat ekstrak cair (y) = b – a
 = 376 gram – 157 gram
 = 219 gram = 219 ml

$$\text{Rendemen} = \frac{y}{x} \times 100\%$$

$$\begin{aligned} \text{Rendemen} &= \frac{219 \text{ ml}}{250 \text{ gram}} \times 100\% \\ &= 87,6\% \end{aligned}$$

Lampiran 3. Perhitungan Penimbangan Bahan

Sediaan hand *wash* dengan bahan aktif ekstrak ampas teh hitam (*Camellia sinensis*) dibuat sebanyak 3 formulasi, yaitu dengan konsentrasi Na₂SO₄ sebesar 5%, 30% dan 45% dengan masing-masing formulasi dibuat sebanyak 100 ml.

Formulasi I

No	Bahan	Konsentrasi (%)	Perhitungan Penimbangan Bahan
1	Ekstrak Ampas Teh Hitam	10	$\frac{10}{100} \times 100 \text{ ml} = 10 \text{ ml}$
2	Na ₂ SO ₄	3	$\frac{3}{100} \times 100 \text{ ml} = 3 \text{ ml}$
3	SLS	6	$\frac{6}{100} \times 100 \text{ ml} = 6 \text{ ml}$
4	STPP	3	$\frac{3}{100} \times 100 \text{ ml} = 3 \text{ ml}$
5	Asam Sitrat	1,5	$\frac{1,5}{100} \times 100 \text{ ml} = 1,5 \text{ ml}$
6	Foam Booster	1,4	$\frac{1,4}{100} \times 100 \text{ ml} = 1,4 \text{ ml}$
7	Bibit Pewangi	0,1	$\frac{0,1}{100} \times 100 \text{ ml} = 0,1 \text{ ml}$
8	Aquadest	ad 100	Add 100 ml

Formulasi II

No	Bahan	Konsentrasi (%)	Perhitungan Penimbangan Bahan
1	Ekstrak Ampas Teh Hitam	10	$\frac{10}{100} \times 100 \text{ ml} = 10 \text{ ml}$
2	Na ₂ SO ₄	6	$\frac{6}{100} \times 100 \text{ ml} = 6 \text{ ml}$
3	SLS	6	$\frac{6}{100} \times 100 \text{ ml} = 6 \text{ ml}$
4	STPP	3	$\frac{3}{100} \times 100 \text{ ml} = 3 \text{ ml}$
5	Asam Sitrat	1,5	$\frac{1,5}{100} \times 100 \text{ ml} = 1,5 \text{ ml}$
6	Foam Booster	1,4	$\frac{1,4}{100} \times 100 \text{ ml} = 1,4 \text{ ml}$
7	Bibit Pewangi	0,1	$\frac{0,1}{100} \times 100 \text{ ml} = 0,1 \text{ ml}$
8	Aquadest	ad 100	Add 100 ml

Formulasi III

No	Bahan	Konsentrasi (%)	Perhitungan Penimbangan Bahan
1	Ekstrak Ampas Teh Hitam	10	$\frac{10}{100} \times 100 \text{ ml} = 10 \text{ ml}$
2	Na ₂ SO ₄	9	$\frac{9}{100} \times 100 \text{ ml} = 9 \text{ ml}$
3	SLS	6	$\frac{6}{100} \times 100 \text{ ml} = 6 \text{ ml}$
4	STPP	3	$\frac{3}{100} \times 100 \text{ ml} = 3 \text{ ml}$
5	Asam Sitrat	1,5	$\frac{1,5}{100} \times 100 \text{ ml} = 1,5 \text{ ml}$
6	Foam Booster	1,4	$\frac{1,4}{100} \times 100 \text{ ml} = 1,4 \text{ ml}$
7	Bibit Pewangi	0,1	$\frac{0,1}{100} \times 100 \text{ ml} = 0,1 \text{ ml}$
8	Aquadest	ad 100	Add 100 ml

Lampiran 4. Perhitungan Uji Tinggi Busa

Formulasi I

Replikasi	Tinggi Busa Awal	Tinggi Busa Akhir	Hasil (%)
1	10,8	6,9	63
2	10,2	6,6	64
3	9,9	6,7	63
Rata-rata (g/ml)	10,3	6,7	63

1. Replikasi 1

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{6,9}{10,8} \times 100\% = 63,88\%\end{aligned}$$

2. Replikasi 2

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{6,6}{10,2} \times 100\% = 64,70\%\end{aligned}$$

3. Replikasi 3

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{6,7}{9,9} \times 100\% = 67,67\%\end{aligned}$$

Formulasi II

Replikasi	Tinggi Busa Awal	Tinggi Busa Akhir	Hasil (%)
1	9,6	7,4	77
2	9,2	6,8	73
3	9,8	7,6	77
Rata-rata (g/ml)	9,5	7,2	75

1. Replikasi 1

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{7,4}{9,6} \times 100\% = 77,08\%\end{aligned}$$

2. Replikasi 2

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{6,8}{9,2} \times 100\% = 73,91\%\end{aligned}$$

3. Replikasi 3

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{7,6}{9,8} \times 100\% = 77,55\%\end{aligned}$$

Formulasi III

Replikasi	Tinggi Busa Awal	Tinggi Busa Akhir	Hasil (%)
1	9,9	8,2	82
2	9,6	7,9	82
3	9,9	7,9	79
Rata-rata (g/ml)	9,8	8,0	81

1. Replikasi 1

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{8,2}{9,9} \times 100\% = 82,82\%\end{aligned}$$

2. Replikasi 2

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{7,9}{9,6} \times 100\% = 82,29\%\end{aligned}$$

3. Replikasi 3

$$\begin{aligned}\text{Uji tinggi busa} &= \frac{\text{Tinggi busa akhir}}{\text{Tinggi busa awal}} \times 100\% \\ &= \frac{7,9}{9,9} \times 100\% = 79,79\%\end{aligned}$$

Lampiran 5. Perhitungan Uji Bobot Jenis

Formulasi I

Replikasi	Pikno Kosong (g) (W ₀)	Pikno + Air (g) (W ₁)	Pikno + Sampel (g) (W ₂)	Volume Piknometer (ml)	ρ sampel (g/ml)
1	23,03	48,51	49,68	25 ml	1,0660
2	23,05	48,51	50,04		1,0796
3	23,04	48,50	50,03		1,0796
Rata-rata (g/ml)	23,04	48,50	49,91		1,0750

1. Replikasi 1

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,51 - 23,03}{25} = 1,0192 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{49,68 - 23,03}{25} = 1,0660 \text{ g/ml}$$

2. Replikasi 2

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,51 - 23,05}{25} = 1,0184 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{50,04 - 23,05}{25} = 1,0796 \text{ g/ml}$$

3. Replikasi 3

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,50 - 23,04}{25} = 1,0184 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{50,03 - 23,04}{25} = 1,0796 \text{ g/ml}$$

Formulasi II

Replikasi	Pikno Kosong (g) (W ₀)	Pikno + Air (g) (W ₁)	Pikno + Sampel (g) (W ₂)	Volume Piknometer (ml)	ρ sampel (g/ml)
1	23,06	48,51	50,36	25 ml	1,0920
2	23,04	48,51	50,18		1,0856
3	23,07	48,50	50,18		1,0844
Rata-rata (g/ml)	23,19	48,50	50,24		1,0873

1. Replikasi 1

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,51 - 23,06}{25} = 1,0192 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{49,68 - 23,06}{25} = 1,0920 \text{ g/ml}$$

2. Replikasi 2

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,51 - 23,04}{25} = 1,0184 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{50,04 - 23,04}{25} = 1,0856 \text{ g/ml}$$

3. Replikasi 3

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,50 - 23,07}{25} = 1,0184 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{50,03 - 23,07}{25} = 1,0844 \text{ g/ml}$$

Formulasi III

Replikasi	Pikno Kosong (g) (W ₀)	Pikno + Air (g) (W ₁)	Pikno + Sampel (g) (W ₂)	Volume Piknometer (ml)	ρ sampel (g/ml)
1	23,17	48,51	49,12	25 ml	1,0380
2	23,22	48,51	49,11		1,0356
3	23,20	48,50	49,11		1,0364
Rata-rata (g/ml)	23,19	48,50	49,11		1,0366

1. Replikasi 1

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,51 - 23,17}{25} = 1,0136 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{49,12 - 23,17}{25} = 1,0380 \text{ g/ml}$$

2. Replikasi 2

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,51 - 23,22}{25} = 1,0116 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{49,11 - 23,22}{25} = 1,0356 \text{ g/ml}$$

3. Replikasi 3

$$\rho_{\text{air}} = \frac{W_1 - W_0}{V_{\text{air}}} = \frac{48,50 - 23,20}{25} = 1,0120 \text{ g/ml}$$

$$\rho_{\text{hand wash}} = \frac{W_2 - W_0}{V_{\text{air}}} = \frac{49,11 - 23,20}{25} = 1,0364 \text{ g/ml}$$

Lampiran 6. Uji Statistik Data

1. Uji Tinggi Busa

ANOVA

Tinggi Busa

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	492.667	2	246.333	85.269	.000
Within Groups	17.333	6	2.889		
Total	510.000	8			

Multiple Comparisons

Dependent Variable: Tinggi Busa

Bonferroni

(I) Perlakuan	(J) Perlakuan	Mean Difference (I-J)	Std. Error	Sig.	95% Confidence Interval	
					Lower Bound	Upper Bound
f1	f2	-12.333*	1.388	.000	-16.90	-7.77
	f3	-17.667*	1.388	.000	-22.23	-13.10
f2	f1	12.333*	1.388	.000	7.77	16.90
	f3	-5.333*	1.388	.026	-9.90	-.77
f3	f1	17.667*	1.388	.000	13.10	22.23
	f2	5.333*	1.388	.026	.77	9.90

*. The mean difference is significant at the 0.05 level.

2. Uji Bobot Jenis

ANOVA

Bobot Jenis

	Sum of Squares	df	Mean Square	F	Sig.
Between Groups	37.556	2	18.778	6.760	.029
Within Groups	16.667	6	2.778		
Total	54.222	8			

Multiple Comparisons

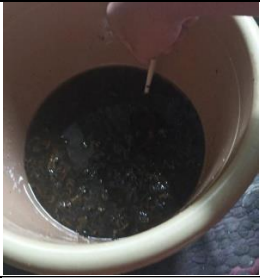
Dependent Variable: Bobot Jenis

Bonferroni

(I) PERLAKUA N	(J) PERLAKUA N	Mean Difference (I-J)	Std. Error	Sig.	95% Confidence Interval	
					Lower Bound	Upper Bound
1	2	-2.667	1.361	.293	-7.14	1.81
	3	2.333	1.361	.412	-2.14	6.81
2	1	2.667	1.361	.293	-1.81	7.14
	3	5.000*	1.361	.031	.53	9.47
3	1	-2.333	1.361	.412	-6.81	2.14
	2	-5.000*	1.361	.031	-9.47	-.53

*. The mean difference is significant at the 0.05 level.

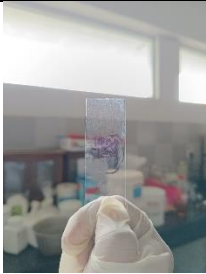
Lampiran 7. Proses Pembuatan Ekstrak

No	Gambar	Keterangan
1		Memasukkan sampel dan pelarut kedalam wadah dengan perbandingan 1:3
2		Wadah ditutup rapat dan didiamkan selama 3 hari pada suhu kamar dalam keadaan kedap cahaya dengan diaduk selama 5 menit setiap 8 jam sekali
3		Menyaring maserat sebanyak 3x replikasi untuk mendapatkan ekstrak
4		Ekstrak ampas hitam (<i>Camellia sinensis</i>) yang telah didapat dimasukkan kedalam wadah dan disimpan untuk pembuatan <i>hand wash</i>

Lampiran 8. Proses Pembuatan Sediaan *Hand Wash*

No	Gambar	Keterangan
1		Proses peparasi alat dan penimbangan bahan
2		Proses pencampuran bahan
3		Proses penambahan ekstrak ampas teh hitam sebagai bahan aktif kedalam bahan yang lain
4		Proses pengemasan

Lampiran 9. Proses Uji Sifat Fisik Sediaan *Hand Wash*

No	Gambar	Keterangan
1		Uji homogenitas
2		Uji pH
3		Uji tinggi busa
4		Uji bobot jenis
5		Uji viskositas

CURICULUM VITAE



BIODATA

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PENDIDIKAN

SD : SDN Kertayasa 03
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SMA : SMA N 03 Tegal
DIII : Politeknik Harapan Bersama
Judul Penelitian : Pengaruh Perbedaan Konsentrasi Sodium Sulfat (Na_2SO_4) Terhadap Sifat Fisik Sediaan *Hand Wash* Dengan Bahan Aktif Ekstrak Ampas Teh Hitam (*Camellia sinensis*)

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